

Formulation Development of Aceclofenac Tablets Employing Starch Phosphate -A New Modified Starch

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Abstract

Purpose: The objective of the study is to prepare, characterize and evaluate starch phosphate, a new modified starch as a carrier in solid dispersions for enhancing the dissolution rate of aceclofenac. The feasibility of formulating solid dispersions of aceclofenac in starch phosphate into compressed tablets with enhanced dissolution rate was also investigated. **Methods:** Starch phosphate was prepared by reacting starch with di-sodium hydrogen orthophosphate anhydrous at elevated temperatures. Solid dispersions of aceclofenac in starch phosphate were prepared by solvent evaporation method employing various weight ratios of drug: starch phosphate such as 2:1(SD-1), 1:1(SD-2), 1:2(SD-3), 1:3(SD-4) and 1:9(SD-5) and were evaluated for dissolution rate and efficiency. Aceclofenac (50 mg) tablets were prepared employing aceclofenac alone and its solid dispersions SD-3 and SD-4 by wet granulation method and were evaluated. **Results and Conclusion:** All the solid dispersions prepared gave rapid and higher dissolution of aceclofenac when compared to pure drug. A 51.89 and 107.03 fold increase in the dissolution rate (K_1) of aceclofenac was observed with solid dispersions SD-4 and SD-5 respectively. The DE_{30} was also increased from 2.50% in the case of aceclofenac pure drug to 69.43% and 79.83% in the case of these solid dispersions. A 4.01 and 18.35 fold increase in the dissolution rate (K_1) was observed with tablet formulations containing solid dispersions SD-3 and SD-4 respectively when compared to plain tablets. Starch phosphate could be used as a carrier to enhance the dissolution rate of aceclofenac from its solid dispersions as well as tablet formulations.

Key words: Starch Phosphate, Aceclofenac, Dissolution Rate, Formulation Development

Introduction

Aceclofenac, a widely prescribed non steroidal anti-inflammatory drug(NSAID) belong to Class II under BCS and exhibit low and variable oral bioavailability due to its poor aqueous solubility. Its oral absorption is dissolution rate limited and it requires enhancement in solubility and dissolution rate for increasing its oral bioavailability. Several techniques[1] such as micronization, cyclodextrin complexation, use of surfactants and solubilizers, solid dispersion in water soluble and dispersible carriers, use of salts, prodrugs and polymorphs which exhibit high solubility, microemulsions and self emulsifying micro and nano disperse systems have been used to enhance the solubility, dissolution rate and bioavailability of poorly soluble drugs. Among the various approaches, solid dispersions in water dispersible excipients is a simple, industrially useful approach for enhancing the solubility, dissolution rate and bioavailability of poorly soluble drugs.

Starch phosphate is one of the modified starches used in the frozen food industry [2,3]. It is produced by phosphorylation of free hydroxyl groups of anhydroglucose units of starch molecule. They are esterified with phosphate reagents. Phosphate reagents for starch phosphate monoester are orthophosphate salts [4]. No reports are available on its use as pharmaceutical excipient. We have earlier reported starch phosphate as an efficient disintegrant in tablet formulations [5].

The objective of the present study is to prepare, characterize and evaluate starch phosphate as a carrier in solid dispersions for enhancing the dissolution rate of aceclofenac. The feasibility of formulating solid dispersions of aceclofenac in starch phosphate into compressed tablets with enhanced dissolution rate was also investigated.

Materials and Methods

Materials

Aceclofenac was gift sample from M/s Dr. Reddys Labs, Hyderabad., Starch phosphate was prepared in the laboratory, Dichloromethane (Qualigens), potato starch (S.D Fine Chemicals), Methanol (S.D Fine Chemicals), lactose, talc, magnesium stearate and acacia were procured from commercial sources.

Methods

Preparation of Starch Phosphate

Starch phosphate was prepared based on the method of Choi *et al*[6] with some modifications. Potato starch (100 g) and di-sodium hydrogen orthophosphate anhydrous (30 g) were suspended in 100 ml of water and continuously stirred for 20 min. This starch slurry was then filtered and the wet starch mixture was conditioned for 12 h at room temperature (28⁰C). To enhance phosphorylation, this mixture was heated in a forced air oven at 130 °C for 3 h. The product obtained was ground and sized.

Characterization of Starch Phosphate

The starch phosphate prepared was evaluated for following

Solubility

Solubility of starch phosphate was tested in water, aqueous buffers of pH 1.2, 4.5, and 7.4 and organic solvents such as alcohol, dichloromethane, chloroform, acetone and petroleum ether.

pH

The pH of a 1% w/v slurry was measured.

Melting Point

Melting point was determined by using melting point apparatus as well as by DSC spectra.

Viscosity

Viscosity of 1% dispersion in water was measured using Ostwald Viscometer.

Swelling Index

Starch phosphate (200 mg) was added to 10 ml of water and light liquid paraffin taken in two different graduated test tubes and mixed. The dispersion in the tubes were allowed to stand for 12 h. The volumes of the sediment in the tubes were recorded. The swelling index of the material was calculated as follows.

$$S.I (\%) = \frac{\text{Volume of sediment in water} - \text{Volume of sediment in light liquid paraffin}}{\text{Volume of sediment in light liquid paraffin}} \times 100$$

Test for gelling property

The gelling property (gelatinization) of the starch and starch phosphate prepared was evaluated by heating a 7% w/v dispersion of each in water at 100⁰C for 30 min.

Moisture absorption

The hygroscopic nature of starch phosphate was evaluated by moisture absorption studies in a closed desiccator at 84% relative humidity and room temperature.

Particle size

Particle size analysis was done by sieving using standard sieves.

Density

Density (g/cc) was determined by liquid displacement method using benzene as liquid.

Bulk density

Bulk density (g/cc) was determined by three tap method in a graduated cylinder.

Angle of repose

Angle of repose was measured by fixed funnel method.

Compressibility index

Compressibility index (CI) was determined by measuring the initial volume (V₀) and final volume (V) after hundred tappings of a sample of starch phosphate in a measuring cylinder. CI was calculated using equation

$$\text{Compressibility index (CI)} = \frac{V_0 - V}{V_0} \times 100$$

Estimation of Aceclofenac

An UV spectrophotometric method based on the measurement of absorbance at 275 nm in phosphate buffer pH 6.8 was used for estimation of aceclofenac. The method obeyed Beer- Lambert's law in the concentration range of 1-10 µm/mL. When the standard drug solution was assayed repeatedly (n=6), the relative

error (accuracy) and coefficient of variation (precision) were found to be 0.60% and 1.0% respectively. No interference from excipients used was observed.

Preparation of Solid Dispersions of Aceclofenac in Starch Phosphate

Solid dispersions of aceclofenac and starch phosphate were prepared in 2:1 (SD-1), 1:1 (SD-2), 1:2 (SD-3), 1:3 (SD-4) and 1:9 (SD-5) ratios of drug: carrier by solvent evaporation method. Aceclofenac (1 g) was dissolved in dichloromethane (10 ml) in a dry mortar to get a clear solution. Starch phosphate (1 g) was then added and mixed. The thick slurry was triturated for 15 min for complete evaporation of dichloromethane and then dried at 55°C until dry. The dried mass was pulverized and sieved through mesh no. 100.

Preparation of Aceclofenac-SD Tablets

Compressed tablets each containing 50 mg of aceclofenac were prepared by wet granulation method employing aceclofenac alone and its solid dispersions (SD-3 and SD-4) in starch phosphate. Lactose was used as diluent to adjust the weight of the tablet to 220 mg. acacia (2%), talc (2%) and magnesium stearate (2%) were incorporated respectively as binder and lubricants.

The tablet granules were prepared by wet granulation method and were compressed into tablets on a Cadmach 16-station rotary tablet punching machine (M/s Cadmach Engineering Co. Pvt. Ltd., Mumbai) using 9 mm concave punches. All the tablets prepared were evaluated for content of active ingredients, hardness, friability, disintegration time and dissolution rate as per official (IP) methods.

Dissolution Rate Study

Dissolution rate of aceclofenac as such and from its solid dispersions and tablets prepared was studied in phosphate buffer pH 6.8 (900 ml) employing USP 8 station Dissolution Rate Test Apparatus (M/s Labindia Disso 8000) with a paddle stirrer at 50 rpm. Aceclofenac or its solid dispersions equivalent of 100 mg of aceclofenac and one tablet containing 50 mg of aceclofenac was used in each test. A temperature $37 \pm 1^\circ\text{C}$ was maintained in each test. Samples of dissolution medium (5 ml) were withdrawn through a filter (0.45μ) at different time intervals and assayed for aceclofenac at 275 nm. For comparison, dissolution of aceclofenac from one commercial brand was also studied. All the dissolution experiments were conducted in triplicate (n=3).

Results and Discussion

Starch phosphate was prepared by reacting starch with di-sodium hydrogen orthophosphate anhydrous at elevated temperatures. The reactions involved are shown in Fig 1. Starch phosphate prepared was found to be white, crystalline, non hygroscopic powder and can easily be ground to different sizes. Powder which passes through mesh no.80 and retained on mesh no.120 was collected. This powder has an average particle size of 152 μm . The starch phosphate prepared was characterised by determining various physical properties. The properties of starch phosphate prepared are summarised in Table 1.

When tested for m.p., it was charred at 210°C . Starch phosphate prepared was insoluble in water, aqueous fluids of acidic and alkaline pH and several organic solvents tested. In water it exhibited good swelling (400%). No gelling/pasting was observed with starch phosphate when its aqueous dispersion was heated at 100°C for 30 min, where as potato starch formed a paste/gel during the above heat treatment. In the micromeritic evaluation, the angle of repose and compressibility index values revealed the excellent flow characteristic of starch phosphate prepared.

As starch phosphate, a chemically modified starch was found to be insoluble in water and has good swelling property without pasting or gelling when heated in water it is considered as a promising carrier for solid dispersions for enhancing the dissolution rate of poorly soluble drugs. Solid dispersions of aceclofenac in starch phosphate were prepared by solvent evaporation method employing various weight ratios of drug: starch phosphate. All the solid dispersions prepared were found to be fine and free flowing powders with an angle of repose in the range $18^\circ - 20^\circ$. Low C.V (< 1.0%) in the percent drug content indicated uniformity of drug content in each batch of solid dispersions prepared.

The dissolution rate of aceclofenac alone and from its solid dispersions was studied in phosphate buffer pH 6.8. All the solid dispersions prepared gave rapid and higher dissolution of aceclofenac when compared to pure drug. The dissolution data were analyzed as per zero order and first order kinetics in each case. The R^2 values were higher in the first order model than in the zero order model indicating that the dissolution of aceclofenac as such and from its solid dispersions followed first order kinetics. The corresponding dissolution rate (K_1) values of various products were estimated. Dissolution Efficiency (DE_{30}) values were calculated as described by Khan *et al* [7]. The dissolution parameters of aceclofenac and its solid dispersions are given in Table 2.

Solid dispersions of aceclofenac showed superior dissolution properties when compared to aceclofenac pure drug. Both dissolution rate (K_1) and DE_{30} values were much higher in the case of solid dispersions when compared to aceclofenac pure drug. The dissolution rate (K_1) and DE_{30} values increased as the proportion of starch phosphate was increased. The number of folds of increase in dissolution rate (K_1) and DE_{30} observed with various solid dispersions are shown in Table 2. A 51.89 and 107.03 fold increase in the dissolution rate (K_1) of aceclofenac was observed with solid dispersions SD-4 and SD-5 respectively.

The DE_{30} was also increased from 2.50% in the case of aceclofenac pure drug to 69.43% and 79.83% in the case of SD-4 and SD-5 respectively. Thus solid dispersions of aceclofenac prepared employing starch phosphate as carrier showed marked enhancement in the dissolution rate (K_1) and DE_{30} of aceclofenac.

The feasibility of formulating aceclofenac solid dispersions in starch phosphate into tablets retaining their rapid and higher dissolution rates was also investigated. Aceclofenac (50 mg) tablets were prepared employing aceclofenac alone and its solid dispersions SD-3 and SD-4 by wet granulation method and were evaluated. All the aceclofenac tablets prepared were found to contain the aceclofenac with in $100\pm 3\%$ of the labelled claim. Hardness of the tablets was in the range 5-8 Kg/sq.cm. Percentage weight loss in the friability test was less than 0.55% in all the cases. Tablets formulated employing solid dispersions disintegrated rapidly with in 1.0 min. Tablets formulated employing aceclofenac pure drug disintegrated within 5-8 min. As such all the aceclofenac tablets prepared were of good quality with regard to drug content, friability, hardness and disintegration time and fulfilled the official (IP) specifications of uncoated tablets.

The dissolution parameters of the prepared tablets are given in Table 3. Dissolution of aceclofenac from all the tablets prepared followed first order kinetics with correlation coefficient ' R^2 ' values > 0.985 . Aceclofenac tablets formulated employing its solid dispersions in starch phosphate (TF2 and TF3) gave rapid and higher dissolution rate and DE_{30} when compared to plain (TF1) and commercial tablets. A 4.01 and 18.35 fold increase in the dissolution rate (K_1) was observed with formulations TF2 and TF3 when compared to formulation TF1. A 2.01 and 9.18 fold increase in the dissolution rate (K_1) was observed with formulations TF2 and TF3 when compared to commercial formulation. Thus solid dispersions of aceclofenac in starch phosphate could be formulated into compressed tablets retaining their fast dissolution characteristics and fulfilling official standards.

Conclusion

Starch phosphate prepared by reacting starch with di-sodium hydrogen orthophosphate anhydrous at elevated temperatures was insoluble in water and has good swelling (400%) property without pasting or gelling when heated in water. Solid dispersions of aceclofenac in starch phosphate prepared by solvent evaporation method employing various weight ratios of drug: starch phosphate gave rapid and higher dissolution of aceclofenac when compared to pure drug. Dissolution followed first order kinetics. A 51.89 and 107.03 fold increase in the dissolution rate (K_1) of aceclofenac was observed with solid dispersions prepared at 1:3 and 1:9 ratios of drug: starch phosphate respectively. The DE_{30} was also increased from 2.50% in the case of aceclofenac pure drug to 69.43% and 79.83% in the case of these solid dispersions. Aceclofenac tablets formulated employing its solid dispersions in starch phosphate also gave rapid and higher dissolution rate and DE_{30} when compared to plain and commercial tablets. A 4.01 and 18.35 fold increase in the dissolution rate (K_1) was observed with tablet formulations containing solid dispersions prepared at 1:2 and 1:3 ratios respectively when compared to plain tablets. Solid dispersions of aceclofenac prepared employing starch phosphate as carrier showed marked enhancement in the dissolution rate (K_1) and DE_{30} of aceclofenac. These solid dispersions could be formulated into compressed tablets retaining their fast dissolution characteristics and fulfilling official (I.P.) standards.

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Table 1
Physical Properties of the Starch Phosphate Prepared

Property	Result
Solubility	Insoluble in all aqueous and organic solvents tested
P ^H (1% w/v aqueous dispersion)	7.25
Melting Point	Charred at 210 ^o c
Viscosity (1% w/v aqueous dispersion)	2.11 cps
Swelling Index	400
Gelling Property	No gelling and the swollen particles of starch phosphate separated from water. Whereas in the case of starch, it was gelatinized and formed gel.
Moisture Absorption	< 4.0 %
Particle Size	152 μm (80/120 mesh)
Density	1.667 g/cc
Bulk Density	0.534 g/cc
Angle of Repose	20.04 ^o
Compressibility Index	11.01 %

Table 2
Dissolution Parameters of the Solid Dispersions of Aceclofenac Prepared Employing Starch Phosphate as a Carrier

Formulation	PD ₁₀ (%)	T ₅₀ (min)	DE ₃₀ (%)	Increase in DE ₃₀ (No of Folds)	K ₁ (min ⁻¹)	Increase in K ₁ (No of Folds)
Aceclofenac	1.84	> 60	2.50	-	0.0013	-
SD-1	22.97	41	27.74	11.09	0.015	11.39
SD-2	56.01	< 5	63.52	25.40	0.054	40.24
SD-3	58.28	< 5	64.84	25.93	0.059	44.76
SD-4	64.11	< 5	69.43	27.78	0.069	51.89
SD-5	76.08	< 5	79.83	31.93	0.143	107.03

Ratio of drug: starch phosphate in solid dispersions: SD-1 (2:1); SD-2 (1:1); SD-3 (1:2); SD-4 (1:3); SD-5 (1:9); PD₁₀ : percent dissolved in 10 min; T₅₀: time for 50 % dissolution; DE₃₀: dissolution efficiency upto 30 min; K₁: first order dissolution rate.

Table 3
Dissolution Parameters of Aceclofenac Tablets Formulated
Employing Aceclofenac alone and its Solid Dispersions in Starch Phosphate

Formulation	PD ₁₀ (%)	T ₅₀ (min)	DE ₃₀ (%)	Increase in DE ₃₀ (No of Folds)	K ₁ (min ⁻¹)	Increase in K ₁ (No of Folds)
TF1	34.79	17	44.82	-	0.035	-
TF2	62.83	7.0	68.29	1.52	0.142	4.01
TF3	87.97	< 5	83.32	1.86	0.650	18.35
Commercial	42.43	12	54.54	1.22	0.071	2.0

TF1: tablets formulated employing aceclofenac alone and using lactose as diluent;
TF2: tablets formulated employing aceclofenac solid dispersion SD-3;
TF3: tablets formulated employing aceclofenac solid dispersion SD-4
PD₁₀ : percent dissolved in 10 min; T₅₀: time for 50 % dissolution; DE₃₀: dissolution efficiency upto 30 min; K₁: first order dissolution rate.

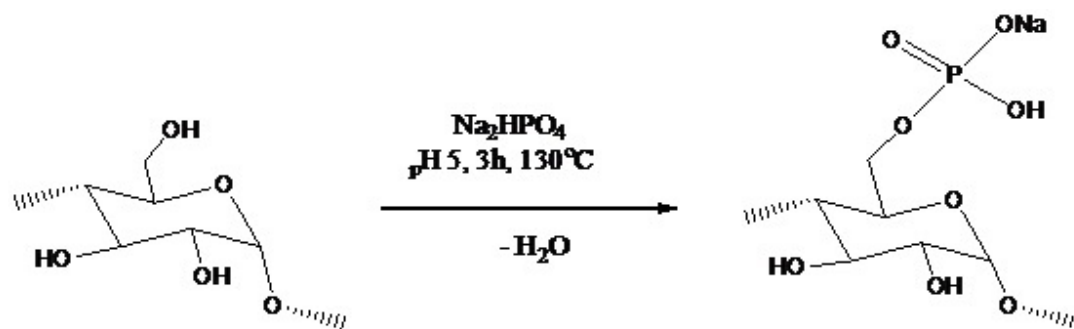


Figure 1: Phosphorification of Potato Starch to Produce Starch Phosphate

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